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 3Express your answer  
 using a chemical  
 symbol. Hg. Part A  
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 oxides in order of

increasing acidity.  
 Rank from least acidic  
 to most acidic. To rank  
 items as equivalent,  
 overlap them. CaO,  
 Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, CO<sub>2</sub>,  
 P<sub>2</sub>O<sub>5</sub>, SO<sub>3</sub>. Part A  
 barium oxide with  
 water Express your  
 answer as a chemical  
 equation. Identify all of  
 the phases in your  
 answer.CHM113  
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Express your answer using a chemical

symbol. Hg. Part A

Arrange the following

oxides in order of

increasing acidity.

Rank from least acidic

to most acidic. To rank

items as equivalent,

overlap them. CaO,

Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, CO<sub>2</sub>,

P<sub>2</sub>O<sub>5</sub>, SO<sub>3</sub>. Part A

barium oxide with

water Express your

answer as a chemical

equation. Identify all of

the phases in your answer.

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