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KAMREN JAMARI

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Al Pa Ca Little, Brown

The periodic table of elements, first encountered by many of us at school, provides an arrangement of the chemical elements, ordered by their atomic number, electron configuration, and recurring chemical properties, and divided into periodic trends. In this Very Short Introduction Eric R. Scerri looks at the trends in properties of elements that led to the construction of the table, and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and, in particular, quantum mechanics, which underlies the behaviour of all of the elements and their compounds. This new edition, publishing in the International Year of the Periodic Table, celebrates the completion of the seventh period of the table, with the ratification and naming of elements 113, 115, 117, and 118 as nihonium, moscovium, tennessine, and oganesson. Eric R. Scerri also incorporates new material on recent advances in our understanding of the origin of the elements, as well as developments concerning group three of the periodic table. ABOUT THE SERIES: The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area. These pocket-sized books are the perfect way to get ahead in a new subject quickly. Our expert authors combine facts, analysis, perspective, new ideas, and enthusiasm

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Chemistry Notebook: Having the right notebook allows you to get things done, and helps you in your goal of being more organized, especially students who are required to taking notes in the class while study at school.Students use notebooks mainly to jot down lecture but not limited to ideas, observations, and things that they find meaningful from their classes.Furthermore, students have a wide variety of preferences for the type of notebook they use. Some are passionate about the size of their notebooks, while others are more concerned that the paper does not allow bleed-through during moments of writing.Here we offer the students with the subject name on the cover from the get go as well as the pictures on the cover uniquely related to that subject. Moreover, we also put in the necessary stuff inside the cover, for instance Periodic Table for Chemistry. Blank line with collage ruled. Size 9.25" x 11.55". 100 Pages.

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A great choice for everyday use during preparation for the GAMSAT, provides plenty of space for notes, lists, musing and more. Features: ✓ Wide-ruled notebook with 100 pages (front and back) for GAMSAT notes, lists, musing and more. ✓ Large dimension - 8.5 x 11 inches - Provides plenty of space for GAMSAT notes. ✓ High quality cover and pages - resists damage from light and air. High quality paper means minimal show-through even when using heavy ink. ✓ Pages are perfect bound - Very durable and intact. ✓ Matte finished cover - Best quality
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Chemists Are Using the Periodic Table of Elements for Their Chemistry Work - This Is the Perfect Notebook for All Chemists

From New York Times bestselling author Sam Kean comes incredible stories of science, history, finance, mythology, the arts, medicine, and more, as told by the Periodic Table. Why did Gandhi hate iodine (I, 53)? How did radium (Ra, 88) nearly ruin Marie Curie's reputation? And why is gallium (Ga, 31) the go-to element for laboratory pranksters? The Periodic Table is a crowning scientific

achievement, but it's also a treasure trove of adventure, betrayal, and obsession. These fascinating tales follow every element on the table as they play out their parts in human history, and in the lives of the (frequently) mad scientists who discovered them. THE DISAPPEARING SPOON masterfully fuses science with the classic lore of invention, investigation, and discovery--from the Big Bang through the end of time. *Though solid at room temperature, gallium is a moldable metal that melts at 84 degrees Fahrenheit. A classic science prank is to mold gallium spoons, serve them with tea, and watch guests recoil as their utensils disappear.

Periodically

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