
Journal Of The American Chemical Society 2011

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Sourcebook on Atomic Energy
Journal of the American Chemical Society
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American Chemical Journal Prentice Hall

Excerpt from The Journal of the American Chemical Society, 1902, Vol. 24: Part II In another experiment grams of the benzene compound of triphenylmethyl took up gram iodine and gave grams of the almost pure amine, while the calculated quantity should have been grams. One recrystallization from ether sufficed

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ammonium oxalate, and sulphuric acid with barium chloride. Then measured the height of a column of the liquid containing the precipitate through which the flame of a common candle was just invisible. I expected only a rude approximation, but to my surprise, I found that between certain limits, an accuracy is attainable equal to that of the ordinary volumetric methods. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works.

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