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The type of bond favored partially depends on the position of the elements in the periodic table. 7 Chemical Formulas and Chemical Compounds CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 5 The Periodic Law jr033.k12.sd.us jr033.k12.sd.us Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative. d. zero. ____ 2. Assessment Chapter Test A CHAPTER 16 REVIEW Reaction Energy SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. For the following examples, state whether the change in entropy favors the forward or reverse reaction: forward reaction a. $\text{HCl(l)} \rightarrow \text{HCl(g)}$ reverse reaction b. $\text{C}_6\text{H}_{12}\text{O}_6\text{(aq)} \leftrightarrow \text{C}_6\text{H}_{12}\text{O}_6\text{(s)}$ forward reaction c. $2\text{NH}_3\text{(g)} \rightarrow \text{N}_2\text{(g)} + 3\text{H}_2\text{(g)}$... 16 Reaction Energy Step-by-step solutions to all your Chemistry homework questions - Slader Chemistry Textbooks :: Free Homework Help and Answers ... Chapter 6 Modern Chemistry Review Answers PDF Online ... Chapter 6 Modern Chemistry Review Answers PDF Online. If you like to read Chapter 6 Modern Chemistry Review Answers PDF Online?? good, means the same to me. did you also know that Chapter 6 Modern Chemistry Review Answers PDF Download is the best sellers book of the year. Chapter 6 Modern Chemistry Review Answers PDF Online ... Chapter 6 Modern Chemistry Review Answers PDF Online. 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Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative. d. zero. ____ 2.

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SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

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