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In the Photoelectric Effect Gizmo, students learn about a very interesting phenomenon. When you shine light on certain metals, the energy from the light can liberate electrons from the surface of the metal. The number and energy of the emitted electrons depends on the frequency and intensity of the light.

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In the Photoelectric Effect Gizmo™, check that the Wavelength is 500 nm, the Intensity is 50%, the Voltage is 0.0 volts, and Potassium is selected. Click Flash the light to send photons of light (green arrows) toward the surface of a metal plate encased in a vacuum tube.

Photoelectric Effect Gizmo Answer Key

Shoot a beam of light at a metal plate in a virtual lab and observe the effect on surface electrons. The type of metal as well as the wavelength and amount of light can be adjusted. An electric field can be created to resist the electrons and measure their initial energies.

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NAME: _____ Student Exploration: Photoelectric Effect Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. Suppose you went bowling, but instead of a bowling ball you rolled a ping pong ball down the alley.

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Gizmo Warm-up The photoelectric effect occurs when tiny packets of light, called photons, knock electrons away from a metal surface. Only photons with enough energy are able to dislodge electrons....

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An emitted electron by the photoelectric effect is called a

photoelectron. Depending on the material and the frequency,
 there may or may not be enough energy to emit an electron. The
 frequency of the light must be above a certain threshold
 frequency,, for the photoelectric effect to take place. Each
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