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particles of an ideal gas are separated by great distances. This implies that the gas

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Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18. Assessment Chapter Test B - Wag & Paws Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20. Assessment Chapter Test B - clarkchargers.org Holt McDougal Modern Chemistry 1 Chapter Test Assessment

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theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

Assessment Chapter Test A - Kettering City School District Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B =  $\sqrt{M_b} / \sqrt{M_a}$

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